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Key indicators

Single-crystal X-ray study T = 150 K Mean σ (C–C) = 0.003 Å Disorder in main residue R factor = 0.038 wR factor = 0.106 Data-to-parameter ratio = 12.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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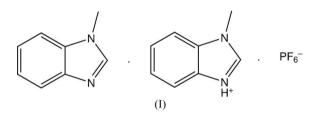
N-Methylbenzimidazole *N*-methylbenzimidazolium hexafluorophosphate

In the title compound, $C_8H_8N_2 \cdot C_8H_9N_2^+ \cdot PF_6^-$, 50% of the *N*-methylbenzimidazole residues are protonated. An N-H···N⁺ hydrogen bond with a $D \cdot \cdot \cdot A$ distance of 2.641 (2) Å is formed. The organic molecules are located on crystallographic mirror planes and the PF_6^- counter-ions are located on crystallographic 2/m sites.

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Comment

During our investigations of ruthenium–bipyridine complexes with DNA model bases (Velders *et al.*, 1999, 2000), we obtained crystals of both *cis*-[Ru(bipyridine)₂(*N*-methylbenzimidazole)₂](PF₆)₂ (Velders *et al.*, 2005) and the title compound, (I), which contains a partly protonated *N*methylbenzimidazole. The Cambridge Structural Database (Version 5.26 of November 2004, Updates 1 and 2; Allen, 2002) reports no other protonated *N*-methylbenzimidazole structures.



The asymmetric unit of (I) contains an *N*-methylbenzimidazole molecule located on a crystallographic mirror plane and a hexafluorophosphate counter-ion positioned on a crystallographic 2/m site. At 2.641 (2) Å from atom N3, a symmetry-related N3 atom is located. The short $N \cdots N$ distance and the electron-density maps strongly suggest the presence of an $[N-H\cdots N]^+$ hydrogen bond, where the H atom displays symmetry-induced disorder (see refinement details and Fig. 2). Atom N3 turns out to be protonated in 50% of the *N*-methylbenzimidazole residues, meaning that the title compound is a co-crystal of neutral *N*-methylbenzimidazole and the hexafluorophosphate salt of protonated *N*-methylbenzimidazole.

The crystal packing of (I) displays layers of partly protonated *N*-methylbenzimidazole and PF_6^- counter-ions (Fig. 3). Due to their location on special positions, the centroids of all residues are located exactly in the *ac* plane. Short C-H···F contacts (Table 2) further stabilize these layers.

Experimental

In a concentrated solution of cis-[Ru(bipyridine)₂(*N*-methylbenzimidazole)₂](PF₆)₂ (0.5 g, 0.5 mol), *N*-methylbenzimidazole (1.1 g,

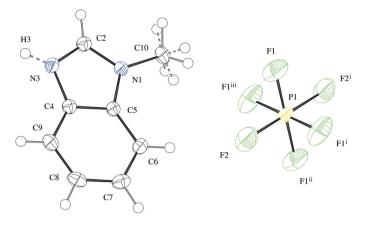


Figure 1

A view of the title compound, showing 50% probability displacement ellipsoids. Atom H3 and the disordered methyl H atoms have siteoccupancy factors of 0.5. [Symmetry codes: (i) -x, y, 1 - z (ii) -x, -y, 1-z (iii) x, -y, z.]

8 mmol) and NH₄PF₆ (3.1 g, 19 mmol) in a water-acetone mixture (1:2 v/v), transparent crystals of the title compound formed after a few weeks at 277 K. The crystalline material was isolated by filtration and washed with water.

 $D_x = 1.520 \text{ Mg m}^{-3}$

Cell parameters from 263

Mo $K\alpha$ radiation

reflections

Block, colourless

 $0.3 \times 0.2 \times 0.2$ mm

 $\theta = 2.0 - 25.0^{\circ}$ $\mu = 0.22~\mathrm{mm}^{-1}$

T = 150 K

Crystal data

 $C_8H_8N_2 \cdot C_8H_9N_2^+ \cdot PF_6^ M_r = 410.31$ Monoclinic, C2/m a = 14.930 (4) Å b = 6.6524 (12) Å c = 9.0261 (12) Å $\beta = 90.214 \ (17)^{\circ}$ V = 896.5 (3) $Å^3$ Z = 2

Data collection

Nonius KappaCCD area-detector 986 reflections with $I > 2\sigma(I)$ diffractometer $R_{\rm int}=0.051$ φ scans and ω scans with κ offset $\theta_{\rm max} = 27.4^\circ$ Absorption correction: none $h = -19 \rightarrow 17$ $k = -7 \rightarrow 8$ 3017 measured reflections $l = -11 \rightarrow 10$ 1072 independent reflections

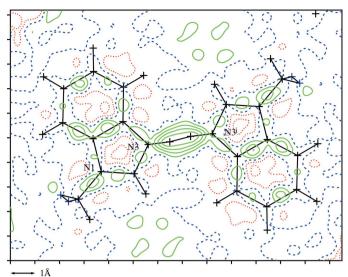
Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0501P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	+ 0.93P]
$wR(F^2) = 0.106$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} < 0.001$
1072 reflections	$\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$
83 parameters	$\Delta \rho_{\min} = -0.32 \text{ e} \text{ Å}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected geometric parameters (Å, °).

N1-C2	1.339 (2)	N3-C2	1.321 (2)
N1-C5	1.380 (2)	N3-C4	1.390 (2)
N1-C10	1.462 (2)		
C2-N1-C10	126.30 (15)	C2-N3-C4	106.44 (15)
C2-N1-C5	107.56 (15)	N1-C2-N3	112.18 (16)
C5-N1-C10	126.13 (15)		





Difference Fourier map of the ac plane, calculated without the contribution of atom H3. Green contours represent positive residual density, red contours represent negative residual density and the blue contours represent the zero level. Contour increment is $0.10 \text{ e} \text{ Å}^{-3}$.

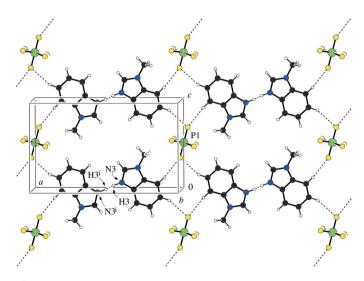


Figure 3

The crystal packing of (I) in the ac plane. Due to their location at special positions, the centroids of all displayed residues are located in the ac plane. The H atom in the $[N-H \cdots N]^+$ hydrogen bond is disordered over positions H3 and H3ⁱ, as indicated in the plot. [Symmetry code: (i) 1 - x, y, -z.]

Table 2

Α , °)
l

	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} N3 - H3 \cdots N3^{i} \\ C6 - H6 \cdots F2 \\ C7 - H7 \cdots F2^{ii} \end{array}$	0.89 (5)	1.76 (5)	2.641 (2)	176 (5)
	0.95	2.52	3.463 (3)	171
	0.95	2.54	3.428 (3)	156

Symmetry codes: (i) -x + 1, y, -z; (ii) -x, y, -z.

The short N3...N3(1 - x, y, -z) distance strongly suggests the presence of a hydrogen bond, either a symmetric $[N \cdots H \cdots N]^+$ bond or an asymmetric $[N-H\cdots N]^+$ hydrogen bond, where the H atom displays symmetry-induced disorder. A difference Fourier map calculated without the contribution of atom H3 displays an elongated area of residual electron density between the two symmetry-related N3 atoms (Fig. 2). This density cannot be resolved into two separate H-atom positions. Refinement of the coordinates of a disordered asymmetrically located H atom, rather then an ordered symmetrically located H atom, resulted in a stable position for the H atom, with an N-H bond length of 0.89 (5) Å. The methyl group of *N*-methylbenzimidazole was refined as a rigid group, allowing for rotation around the N-C bond and displaying disorder over the crystallographic mirror plane in which N-C is located. All other H atoms were introduced in calculated positions, riding on their carrier atoms, with C-H = 0.95-0.98 Å. The constraint $U_{iso}(H) = 1.2 U_{eq}(carrier)$ was applied for all H atoms.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *DENZO* (Otwinowski & Minor, 1997); data reduction: *DENZO*; program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1985). SHELXS86. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Velders, A. H., Hotze, A. C. G., Haasnoot, J. G. & Reedijk, J. (1999). Inorg. Chem. 38, 2762–2763.
- Velders, A. H., Hotze, A. C. G., van Albada, G. A., Haasnoot, J. G. & Reedijk, J. (2000). *Inorg. Chem.* **39**, 4073–4080.
- Velders, A. H., Kooijman, H. & Spek, A. L. (2005). In preparation.