Acta Crystallographica Section E Structure Reports Online ISSN 1600-5368 Editors: W. Clegg and D. G. Watson

Dimethyl fumarate

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Huub Kooijman *et al.* • C₆H₈O₄

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 150 KMean σ (C–C) = 0.002 Å R factor = 0.045 wR factor = 0.122 Data-to-parameter ratio = 10.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Dimethyl fumarate

Dimethyl fumarate, $C_6H_8O_4$, crystallizes in the fully stretched planar conformation. The molecule is located on a crystallographic inversion centre and has local symmetry 2/m. The fumarate molecules aggregate in layers coinciding with the (11) lattice planes.

Comment

Dimethyl fumarate, (I), is frequently used as a coordinating ligand. The January 2004 update of the Cambridge Structural Database (CSD; Allen, 2002) contains 62 crystal structures of compounds with dimethyl fumarate acting as a ligand coordinated to a metal atom, mostly Ru and Pt. Knaggs & Lonsdale (1942) reported the unit-cell parameters of the crystal structure of the isolated molecule. Up until now, the full crystal structure determination of this compound has not been reported. The only other diester of fumaric acid reported in the CSD is diphenyl fumarate (Albano *et al.*, 2003).



Dimethyl fumarate adopts the fully stretched planar conformation in the crystal structure reported in this paper (Fig. 1). The molecule is located on a crystallographic inversion centre and has local symmetry 2/m, in contrast with diphenyl fumarate, which has neither crystallographic nor local symmetry. The intermolecular contact $C3-H3B\cdots O2(1+x)$, 1 + y, z) joins the molecules into chains running in the [110] direction (Fig. 2). The H···O distance for this contact is 2.48 (3) Å, *i.e.* 0.24 Å shorter then the sum of the van der Waals radii. The orientations of groups within these chains suggest the presence of a close contact $C1-H1\cdots O1(1-x)$, 1 - y, 1 - z). With an H···O distance of 2.65 (2) Å, *i.e.* 0.07 Å shorter then the sum of the van der Waals radii, this contact is less pronounced. The chains of fumarate molecules are arranged into layers in such a way that all non-H atoms are more or less located on one of the $(1\overline{1}1)$ lattice planes (Fig. 3).



Figure 1

View of the title compound, with the atom-numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 50% probability level. [Symmetry code: (i) -x, -y, -z.]

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DOI: 10.1107/S1600536804010177

Received 23 April 2004 Accepted 27 April 2004 Online 30 April 2004

Experimental

Crystals of the title compound were grown from a solution containing [platinum(dimethylfumarate)(norbornene)₂] and excess dimethyl fumarate in ether-pentane (2:1) at 253 K (Sprengers et al., 2004). The dimethyl fumarate crystals in this batch were of highly irregular shape, but displayed reasonable extinction effects under crossed polarization filters.

542 reflections with $I > 2\sigma(I)$

 $R_{\rm int} = 0.037$ $\theta_{\rm max} = 25.2^{\circ}$ $h = -4 \rightarrow 3$ $k = -6 \rightarrow 6$ $l = -9 \rightarrow 10$

Crystal data

$C_6H_8O_4$	Z = 1
$M_r = 144.12$	$D_x = 1.429 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 3.8680 (14) Å	Cell parameters from 530
b = 5.643 (2) Å	reflections
c = 8.364 (3) Å	$\theta = 2.0-25.0^{\circ}$
$\alpha = 100.843 \ (16)^{\circ}$	$\mu = 0.12 \text{ mm}^{-1}$
$\beta = 100.289 \ (18)^{\circ}$	T = 150 K
$\gamma = 105.706 \ (14)^{\circ}$	Irregular block, colourless
$V = 167.44 (11) \text{ Å}^3$	$0.3 \times 0.2 \times 0.2 \text{ mm}$

Data collection

Nonius KappaCCD area-detector
diffractometer
φ scans, and ω scans with κ offsets
Absorption correction: none
1315 measured reflections
597 independent reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0721P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	+ 0.03P]
$wR(F^2) = 0.122$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.11	$(\Delta/\sigma)_{\rm max} < 0.001$
597 reflections	$\Delta \rho_{\rm max} = 0.24 \ {\rm e} \ {\rm \AA}^{-3}$
58 parameters	$\Delta \rho_{\rm min} = -0.26 \text{ e } \text{\AA}^{-3}$
Only coordinates of H atoms	
refined	

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$\begin{array}{c} C1 - H1 \cdots O1^{i} \\ C3 - H3B \cdots O2^{ii} \end{array}$	0.96 (2)	2.65 (2)	3.603 (2)	173.2 (14)
	0.95 (3)	2.48 (3)	3.416 (2)	167 (2)

Symmetry codes: (i) 1 - x, 1 - y, -z; (ii) 1 + x, 1 + y, z.

The crystals diffracted poorly. Broad fuzzy reflections were observed and the mosaicity of the crystal refined to approximately 3.0°. H atoms were located in a difference Fourier map and their coordinates were refined [C-H = 0.94 (2)-0.96 (2) Å]. The isotropic displacement parameters of the H atoms were set at 1.5 or 1.2 times the equivalent isotropic displacement parameter of the carrier atom for methyl and alkene H atoms, respectively.

Data collection: COLLECT (Nonius, 1998); cell refinement: DENZO (Otwinowski & Minor, 1997); data reduction: DENZO; program(s) used to solve structure: SHELXS86 (Sheldrick, 1985); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: PLATON.



Figure 2

The title compound forms chains joined by close C-H···O contacts, running in the [110] direction.



Arrangement of chains of dimethyl fumarate molecules into layers coinciding with the $(1\overline{1}1)$ lattice planes.

This work was supported in part (ALS) by the Council for the Chemical Sciences of the Netherlands Organization for Scientific Research (CW-NWO). The National Research School Combination Catalysis (project No. 2000-14) is gratefully acknowledged for supporting this work in part (JWS, MJA and CJE).

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